



ULTIMATE TEST SERIES NEET -2020

TEST-07 SOLUTION

Test Date :20-03-2020

[PHYSICS]

1. Potential at earthed conductor becomes zero.

2. Due to slab.

$$C \rightarrow KC, \quad E = \frac{1}{2} CV^2$$

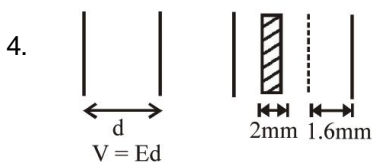
$$V \rightarrow V/K, \quad E = E/K$$

$$Q = CV = \text{constant}$$

V → Decrease, Energy decrease.

Q → Remain constant

3. Potential at earthed conductor becomes zero and by induction charge will not remain uniform.



$$Ed = V = (d + 1.6 - 2)E + \frac{E}{K} \quad .2$$

$$\frac{2}{K} = \frac{4}{10} \quad \boxed{K=5}$$

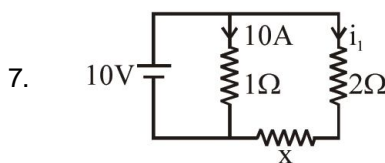
5. Flux donot depend upon shape

$$w = \vec{F} \cdot \vec{d}$$

$$= q_0 \vec{E} \cdot \vec{d}$$

$$= q_0 (E_0 \hat{i}) \cdot (a\hat{i} - a\hat{j})$$

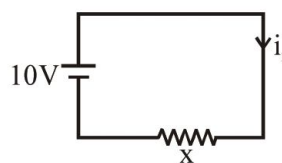
$$= q_0 E_0 a$$



(when switch is opened)

$$10 \times 1 = i_1 \times (2 + x)$$

$$i_1 = \frac{10}{2+x} \quad \dots(1)$$



$$i_2 = \frac{10}{x}$$

$$\therefore i_2 = 2i_1$$

$$\frac{10}{x} = 2 \left( \frac{10}{2+x} \right)$$

$$x = 2\Omega$$

8.  $\leftarrow | \rightarrow$  T.P.D = E - I r  $\Rightarrow 12 = E - 2r \dots(1)$

$\rightarrow | \rightarrow$  T.P.D = E + I r  $\Rightarrow 15 = E + 3r \dots(2)$

Solve eq<sup>n</sup> (1) & (2)

9. When wire is stretched

Volume = constant

$Al = \pi r^2 l = \text{constant}$

$$l \propto \frac{1}{r^2}$$

Resistance  $R = \frac{\rho l}{A}$  so  $R \propto \frac{1}{r^4}$

10.  $V = x\ell$

$IR = x \times 100$

$I(R + R) = x \times \ell'$

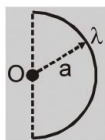
$$\frac{1}{2} = \frac{100}{\ell'}$$

$\Rightarrow \ell' = 200 \text{ cm}$

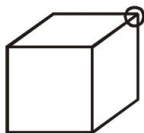
11.  $E_{\text{arc}} = \frac{2k\lambda}{a} \sin(\alpha/2)$

for this figure  $\alpha = 180^\circ$

hence  $E = \frac{2k\lambda}{a} = \frac{\lambda}{2\pi\epsilon_0 a}$



12. Faces which are related to the corner will have zero flux ( $E \perp A$ ).



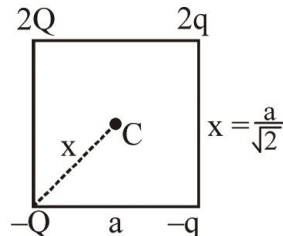
13. Deviation  $y = \frac{1}{2} \left( \frac{qE}{m} \right) \left( \frac{x^2}{v^2} \right)$

But putting the values we get  $y = 1.76 \text{ mm}$

14. As charge moves towards 'A' more number of field lines will be related with 'A' hence  $\phi_B$  will decrease.

15.  $V_c = \frac{k(-Q)}{x} + \frac{k(-q)}{x} + \frac{k(2q)}{x} + \frac{k(2Q)}{x}$

As per question  $V_c = 0$



$$a + q = 2q + 2Q$$

$$Q = -q$$

16. The given point of observation for the given sphere is inside. Hence field inside the conducting sphere is zero.

17.  $F = -\frac{dU}{dx}$

$$F = \vec{p} \cdot \frac{d\vec{E}}{dx}$$

as  $\theta = 90$

$F = 0$

$\theta \neq 90$

$F \neq 0$

If dipole is aligned with EF lines hence  $\tau = 0$

18.  $\left( \text{Energy loss during the process} \right) = \frac{c_1 c_2}{2(c_1 + c_2)} (V_1 - V_2)^2$

for no loss of energy  $V_1 - V_2 = 0 \Rightarrow V_1 = V_2$

If  $Q_1 R_2 \neq Q_2 R_1$  then there is always a loss in energy of the system hence option (4) is correct.

19. Capacity when dielectric is filled partially is

$$c' = \frac{\epsilon_0 A}{d - t \left( 1 - \frac{1}{\epsilon_r} \right)}$$

Hence  $c = \frac{\epsilon_0 A}{d}$

and  $c' = \frac{\epsilon_0 A}{d - d/2 \left(1 - \frac{1}{\infty}\right)} = \frac{2 \epsilon_0 A}{d} = 2c$

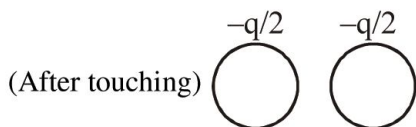
$\frac{c'}{c} = 2$  for copper  $\epsilon_r = \infty$

20.  $C_{PR} = C/3 + C/2 = \frac{5}{6}C$

$C_{PQ} = \frac{C}{4} + C = \frac{5}{4}C$

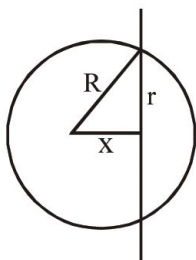
$C_{PR} = C_{PQ} = \frac{1}{6} : \frac{1}{4} = 2 : 3$

21.  $F = \frac{K(q)(2q)}{r^2} = \frac{2Kq^2}{r^2}$  (before touching)



$F' = \frac{K(q/2)(q/2)}{r^2} = \frac{kq^2}{4r^2} \Rightarrow F' = F/8$

22. Charge enclosed  $q_{en} = \sigma\pi(R^2 - x^2)$



Here  $r = \sqrt{R^2 - x^2}$

$\phi_{sphere} = \frac{\sigma\pi(R^2 - x^2)}{\epsilon_0}$

23. Work done = change in energy

$eEd = \frac{1}{2}m(V \cos 60^\circ)^2 = \frac{1}{4} \left( \frac{1}{2}mV^2 \right)$

but  $\frac{1}{2}mV^2 = K$

$E = \frac{K}{4ed}$

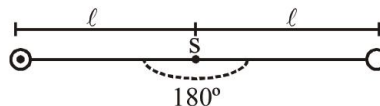
24. for surface points, net charge = 0  
hence potential at surface = 0

(a)  $\phi = \frac{q_{in}}{\epsilon_0}$  but  $q_{in} = 0$   
Hence  $\phi = 0$

(d) Concentric circle which lies on equatorial plane will have zero potential.

25. Potential difference does not depend upon charge of outer sphere.

26. In absence of gravity only electrostatic force will work.

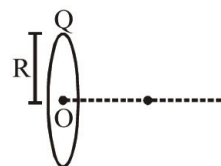


$T = \frac{K(Q)(Q)}{(2l)^2}$  and angle between strings =  $180^\circ$

27.  $TE_i = TE_f$

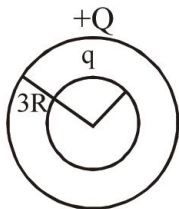
$\frac{KQq}{R} = \frac{1}{2}mv^2$

$v = \sqrt{\frac{2KQq}{mR}}$



28. If inner sphere is earthed then its potential will be zero.

Let charge on inner sphere is  $q$ .

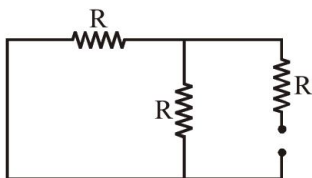


$$V_{\text{inner}} = \frac{KQ}{3R} + \frac{Kq}{R} = 0$$

$$q = -\frac{Q}{3}$$

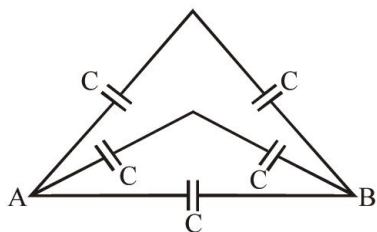
29. If cell is connected potential difference will remain same but capacitance increases so charge will increase.

30. Net resistance across capacitor is  $\frac{3R}{2}$



$$\tau = R_N C = \frac{3}{2} RC$$

31. Balanced (ws bridge)



$$C_N = 2C$$

32.  $V - V_{(0,0,0)} = \int_0^1 E_x dx + \int_0^1 E_y dy$

$$V - 2 = [x^2]_0^1 + \frac{1}{3}[y^3]_0^1$$

$$V = 2 + 1 + \frac{1}{3} = \frac{10}{3} \text{ volt}$$

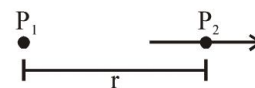
33.  $x = \frac{r\sqrt{q_1}}{\sqrt{q_1} + \sqrt{q_2}}$  x is distance from  $q_1$ .

$$x = \frac{30\sqrt{e}}{\sqrt{e} + \sqrt{9e}} = \frac{30}{3} = 10 \text{ cm from } e$$

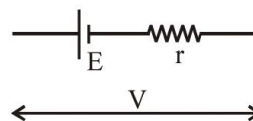
34.  $F = P_2 \frac{d}{dr} \left( \frac{2KP_1}{r^3} \right)$

$$F = 2KP_1 P_2 \frac{d}{dr} r^{-3}$$

$$F = -6 \frac{KP_1 P_2}{r^4}$$



35. When circuit is open  $V = E$   
 $\therefore E = 2.2 \text{ V}$



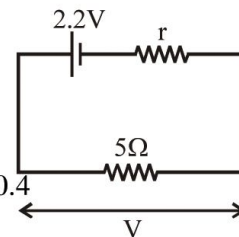
$$V = IR = 1.8$$

$$\therefore I = \frac{1.8}{5} = 0.36 \text{ Amp.}$$

$$V = E - Ir$$

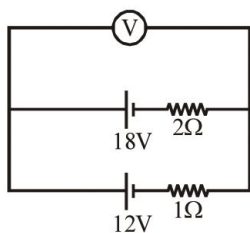
$$Ir = E - V = 2.2 - 1.8 = 0.4$$

$$r = \frac{0.4}{I} = \frac{0.4}{0.36} = \frac{10}{9} \Omega$$



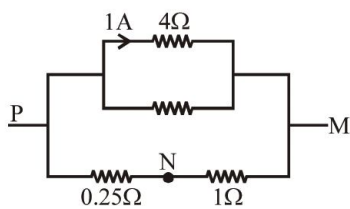
36. 
$$E_{\text{net}} = \frac{E_1 r_2 + E_2 r_1}{r_1 + r_2}$$

$$= \frac{(18 \times 1) + (12 \times 2)}{(2 + 1)}$$



$$E_{\text{net}} = \frac{18 + 24}{3} = \frac{42}{3} = 14 \text{ volt}$$

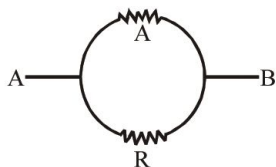
37.  $V_{\text{pm}} = 4V$



$$\therefore V_{\text{MN}} = \frac{1}{1 + 0.25} \times 4$$

$$= 3.2 \text{ volt}$$

38.  $R = \frac{(2\pi \times 0.1)12}{2}$



$R = 1.2 \pi$

$$\therefore R_{\text{AB}} = R/2 = \frac{1.2\pi}{2} = 0.6\pi \Omega$$

39.  $I_g \cdot G = (I - I_g) \cdot S$

$$\therefore S = \frac{I_g \cdot G}{(I - I_g)}$$

Given  $I = 5A$ ,  $I_g = 1 \text{ Amp.}$ ,  $G = 60 \text{ M}$

$$\therefore S = \frac{60 \times 1}{4} = 15 \Omega \text{ in parallel}$$

40. Power dissipated in  $5\Omega$

$$= i^2 R$$

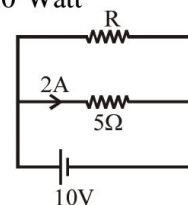
$$= (2)^2 \times 5 = 20 \text{ W}$$

$$\therefore \text{Power dissipated in } R = 30 - 20$$

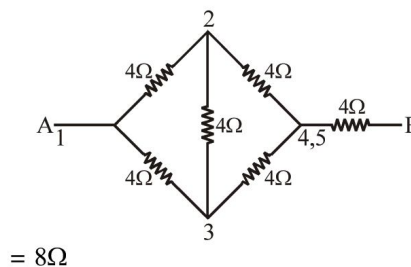
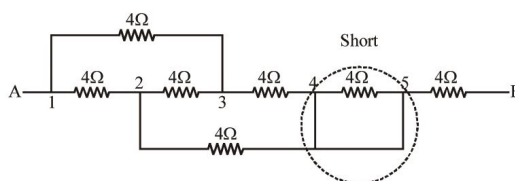
$$= 10 \text{ Watt}$$

$$\therefore 10 = \frac{(10)^2}{R}$$

$$R = \frac{100}{10} = 10 \Omega$$

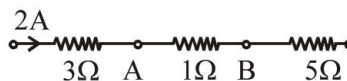


41.



$= 8\Omega$

42. Given circuit can be redrawn as



$$V_A - V_B = IR = 2 \times 1 = 2V$$

43. Current in  $6\Omega$   $I = \sqrt{\frac{P}{R}} = \sqrt{\frac{6}{6}} = 1$

$$I = \frac{12}{6 + \frac{8P}{8 + R}}$$

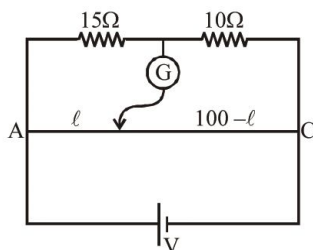
$$1 = \frac{12}{48 + 14R} (8 + R)$$

$$R = 24\Omega$$

$$44. \frac{R_1}{R_2} = \frac{\ell}{100-\ell}$$

$$\Rightarrow \frac{15}{10} = \frac{\ell}{100-\ell}$$

$$\Rightarrow \ell = 60 \text{ cm}$$



$$45. \text{T.P.D.} = E - Ir = E - \left(\frac{E}{r}\right)r = 0$$

(here  $4\Omega$  is short circuited so it is use less)

## [CHEMISTRY]

$$46. \Delta S = \ominus \text{ve}, \Delta H = \ominus \text{ve}$$

47.  $\text{Fe}^{+3}$  ion, according to Hardy – Schulze law.

$$48. R = K[\text{RCI}]^1 [\text{H}_2\text{O}]^0$$

$$49. E_{\text{cell}} = E_{\text{cell}}^0 = \frac{0.0591}{2} \log_{10} \frac{[\text{Zn}^{+2}]}{[\text{Cu}^{+2}]}$$

$$50. kt = 2.303 \log \frac{a}{a-x}$$

51. Only C reduces  $\text{H}^+$  therefore element A, B and D are below in E.C.S. than hydrogen  
 $\Rightarrow$  A reduces only ion of D therefore it's position in E.C.S. is above than D.  
 $\Rightarrow$  Increasing order of SRP  $\rightarrow \text{C} < \text{H} < \text{B} < \text{A} < \text{D}$

52. In this reaction :  
 Intermediates  $\Rightarrow \text{N}_2\text{O}_2$  and  $\text{N}_2\text{O}$

$$53. \text{specific resistance } (k = \frac{1}{\rho} = \frac{1}{R} \times \frac{1}{a})$$

$$\pi_m = \frac{k \times 1000}{\text{molarity}}$$

$$54. r = k(\text{A}_2)^x (\text{B})^y$$

$$55. \alpha = \frac{\wedge_m}{\wedge_m^\infty}$$

$$K_a = \frac{C\alpha^2}{1-\alpha}$$

56. Conductivity of strong electrolyte increasing on dilution due to increase in inter ionic distance.

57. Mg is obtain by electrolysis of molten solution of  $\text{MgCl}_2$ .

$$58. kt_{1/4} = 2.303 \log_{10} \frac{a}{3a/4}$$

$$59. \text{SO}_3^{2-} \Rightarrow 1(x) + 3(-2) = -2 \therefore x = +4$$

$$\text{S}_2\text{O}_4^{2-} \Rightarrow 2(x) + 4(-2) = -2 \therefore x = +3$$

$$\text{S}_2\text{O}_6^{2-} \Rightarrow 2(x) + 6(-2) = -2 \therefore x = +5$$

$$60. t_{1/2} = \frac{0.693}{K}$$

$$\log \frac{a}{a-x} = \frac{kt}{2.303}$$

61. Central atom nitrogen (O.N. = +3) present in intermediate oxidation state so it can act as oxidant as well as reductant.

$$62. \text{R.O.R.} = -\frac{d[\text{A}]}{dt} = -\frac{d[\text{B}]}{dt} = \frac{1}{2} \frac{d[\text{C}]}{dt} = \frac{d[\text{D}]}{dt}$$

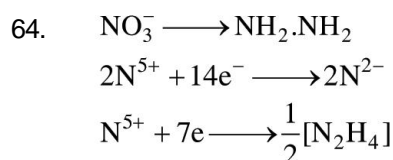
$$\text{R.O.D. of A} = -\frac{d[\text{A}]}{dt}$$

$$\text{R.O.D. of B} = -\frac{d[\text{B}]}{dt}$$

$$63. R = k[\text{NO}_2]^1$$

order of reaction = 1

$$t_{1/2} = \frac{0.693}{k}$$



65.  $K = \frac{1}{R} \left( \frac{\ell}{A} \right)$   
 $\left( \frac{\ell}{A} \right) = 0.013 \times 300 \text{ cm}^{-1}$
66. fact (refer theory of catalyst)
67.  $w = \frac{E}{F} \times It$   
 $24 \times 5 = \frac{27/3}{96500} \times 9650 \times t \times \frac{90}{100}$
68. Only C is correct rest are incorrect
69. fact
70. In peptization freshly prepared precipitate is converts into sol.
71. On iron surface iron itself act as anode an get oxidised an  $O_2$  in water get reduced
72.  $\ln k = \ln A - \frac{E_a}{R \times T}$
73.  $E^\circ = \frac{0.0591}{2} \log_{10} \text{Keq.}$   
 $0.2955 = \frac{0.0591}{2} \log \text{Keq.} \Rightarrow \text{Keq.} = 10^{10}$
74.  $\pi = i CRT$   $i = 1$
75. -
76. B is present in octahedral void
77.  $T_f^\circ - T_f = i \times k_f \times m$
78. fact
79.  $d = \frac{z \times m}{a^3 \times N_A}$
80.  $\Delta T_f = i \times k_f \times m$
81.  $r = \frac{a}{2\sqrt{2}}$
82.  $Zn + H_2SO_4 \rightarrow ZnSO_4 + H_2$   
 65 g of Zn evolve 22400 ml  $H_2$   
 therefore  $0.5 \times 10^{-2}$  mol  $H_2$  evolved by 0.32g Zn
83. More the value of vant'Hoff facto more will be boiling point
84.  $m = \frac{x_B 1000}{x_A \times M_A}$
85. 50 ml of 5.6 % w/v – KOH contain 4.91 mol of KOH & mol 5.6% w/v HCl contain 7.67 mol of HCL. SO acid is more and base neutralises completely
86.  $r_{\text{new}} = r_{\text{old}} \times \mu^{\Delta T/10}$   
 $\Delta T = 55 - 25 = 30$
87.  $E = E^\circ - \frac{0.059}{2} \log \frac{[Cu^{2+}]}{[Ag^+]^2}$   
 As  $[Ag]^+$  increase twice  $\frac{[Cu^{2+}]}{[Ag^+]^2}$  become  $\frac{1}{4}$  and  
 E shows more change and on halving the  $[Cu^{2+}]$   
 $\frac{[Cu^{2+}]}{[Ag^+]^2} = \frac{1}{2}$  of intial value
88.  $W = \frac{E}{96500} \times Q$   
 $710 = \frac{35.5}{96500} \times Q$   
 $Q = 1930000 = 1.93 \times 10^6 \text{ C}$
89. Physisorption is multilayered
90. Fact